ACADEMIE ROYALE DE BELGIQUE-COLLEGE BELGIQUE 10 novembre 2016

#### LES BIOMINERAUX MICROBIENS DES GISEMENTS TERRESTRES À L'EXOBIOLOGIE

PARTIE II POURQUOI DONC LES MARBRES ROUGES SONT-ILS ROUGES?

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#### STROMATOLITHES ET TAPIS MICROBIENS PRECAMBRIENS (CONGO-BRAZZAVILLE)



WHY IS 'RED MARBLE' RED : COULD FE-ISOTOPES SHED LIGHT ON THIS QUESTION THROUGH THE STUDY OF THE ITALIAN AMMONITICO ROSSO AND RECENT ORGANISMS?

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## **Red** limestones are rare but precious ...

# THE PROBLEM is their red pigmentation



#### OF WHAT CONSISTS THE RED PIGMENTATION ?



# Our studied Red limestones ... PALEOZOIC and MESOZOIC



**OF WHAT CONSISTS THE RED PIGMENTATION ?** A simple solution is just to ignore the problem!

The Ammonitico Rosso Symposium (1991) The thick abstract book with its appropriate SCARLET cover deals with sedimentology, palaeontology, diagenesis but nothing on the origin of the colour of the Ammonitico Rosso !





#### Researches in our laboratory, ULB/1989

... 'Frasnian mud mounds of the Dinant basin are bioconstructions built during a regressive phase that passes from an aphotic to an euphotic zone, and through the dysphotic level where the red marbles are concentrated' ...

## BEAUCHATEAU QUARRY



° The base of the bioconstructions is in very calm environments under the SWB, and then the mounds pass into the FWWB where sedimentation ends

° No signs of subaerial exposure as observed in modern reefs



69 reported 'red 'mud mounds Severe eustatic sea level rises High vertical facies differenciation High content in microaerophilitic iron bacteria (in *stromatactis* cavities) Submicronic hematite hexagonal plates dispersed in the matrix Fe<sub>2</sub>O<sub>3</sub> : average 2% (max 5%) Diam 100-250'm X H 30-80m Only 30m (upper part) are visible on ±60m



BEAUCHATEAU QUARRY near Senzeilles Philippeville Massif BELGIUM

#### Frasnian 'F2j' mud mound

#### Griottes + 'stromatactis' (1880)

#### The red-pink-grey color (succession) is 'ECOLOGICAL'



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#### The red-pink-grey colour (succession) is 'ECOLOGICAL'





#### | : recurrence

- 7-8-9 : FWWB, cyanobacterial, algal (green algae)-coral-peloid wackestones-packstones
- 5-6 : progressive biotic enrichment (stromatoporoids, corals...)
- 4 : SWB, oligophotic environment (corals, crinoids, stromatactis)
- 3 : <u>iron bacteria</u>-sponge in a quiet aphotic/hypoxic environment, <u>stromatactis</u>
- 2: shale and carbonates with brachiopods, corals, crinoids, sponges
- I : shale with poor fauna, mainly <u>sponges</u> (substrate)





# GEOLOGY

Iron bacteria and iron coccoids (hematite in the sheaths) Granular sparite, internal sediment Relicts of iron bacteria in a slitghtly microsparitized matrix (griotte )

# BIOLOGY

Present day iron bacteria (Sphaerotilus-Leptothrix group) Small stream in Brussels



'Perfectly' fossilized during cementation of the cavity (stromatactis) Strongly altered-broken during 'recrystallization' (microsparitization) of the micrite matrix Red color due to ruthenium (interferential contrast) Beggiatoa sp. (microaerobic)

... how can we explain the red colour that made the stone so rare?

#### <u>XVIII<sup>th</sup>-XIX<sup>th</sup> centuries</u>: red = iron (Delhaye, 1908)

° the iron is detrital (Reijers, 1985), transported from the continent, then mixed with the carbonate matrix during sedimentation ...

° its concentration and degree of oxidation produce colour variations (reddish)

LATER ON (1964-1988): a relation between ferruginization/palaeogeography/climate is the fashion. Washed equatorial laterite soils provide great quantities of iron oxides... thus the red limestones are used as palaeoclimatic indicators!

# Oxygenation degree ? (in non clastic rocks)

Red limestones are found in oxidized facies Green limestones indicate reducing conditions

both indicate shallow waters

BUT ...

Red silicified limestones (lydites) indicate deep environments with minimum amounts of iron and oxygen ! <u>and</u> they are not reduced...

# Our first basic observation

#### The Fe content of red limestones of biotic origin is low



# ... therefore this content is not responsible of the coloration

# Our second basic observation

# Sedimentation is in a NORMAL OPEN MARINE facies

Red limestones are formed in calm environments, with low levels of oxygen

# The colour <u>is not</u> necessarily linked to shallow water marine environments where oxygen is abundant





#### and its biofilms...





OUTER LAYER: BACTERIAL MAT (LIVING FILAMENTOUS BACTERIA MAINLY IN DYS /ANAEROBIC CONDITIONS) *∂* –*Proteobacteria* (Desulfonema G.), *Bacteroidetes*, *Firmicutes* + *Thiothrix*?



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Gillan & De Ridder 2001. Chemical Geology 177:371-379

# Filamentous iron bacteria from the shell of *M. ferruginosa* (base of outer layer)



Bacterial filaments on the shell (photonic microscopy) Bacterial filament - detail TEM (F: Fe)





#### LIVING MODELS



#### LIVING MODELS



Rust-coloured and ferric iron encrusted biofilm (3mm thick)

#### MICROBIAL MAT WITH THREE DISTINCT LAYERS

- 1. Outer layer Filamentous and coccoid bacteria, also protozoa
- 2. Intermediate layer <u>Microbial AND mineral</u> (heavily ferric iron-encrusted filamentous bacteria and protozoa
- 3. Inner layer Essentially mineral (ferric iron deposits),

no or rare living microorganisms

#### GENESIS OF THE MINERAL-MICROBIAL MAT

- I. Ferric iron *deposition* within bacterial sheaths in the outer layer
- 2. <u>Release and accumulation</u> of heavily ferric iron-encrustated sheaths after lysis of the bacteria in the intermediary layer
- 3. <u>Degradation</u> of bacterial sheaths and *accumulation* of ferric iron minerals in the inner layer

ferric iron deposits = 0.05-1  $\mu$ m (granules, amorphous oxyhydroxide 'gel' with phosphate)

#### = PASSIVE PROCESS => BIOMINERALIZATION-INDUCED and BIOACCUMULATION (microbes dead or alive)

From PhD thesis C. DE RIDDER 1986 and D. GILLAN, 1999, Marine Biology Dept, ULB and several papers in Journals of Microbiology...) ... BACK TO GEOLOGY I OBSERVED MICROFACIES OF DIFFERENT AGES AND LOCALITIES



#### OF WHAT CONSISTS THE RED PIGMENTATION?



- ° infillings of original fossil cavities
- ° calcite replacement of dissolved echinoderm plates
- ° infillings of bioperforations
- ° bacterial/fungal filaments
- ° 'hedgehogs' and 'erythrospheres'
- ° massive hematite/goethite coating around microfossils
- ° simple or multiple biofilms
- microstromatolites (exogens ou endogens, crenulated or not...)
- ° oncolites
- ..... non exhaustive ....



inframicrometric hematite crystals coating bacterial filaments 0.5 µm



benthic bacterial mats up to 20%

Préat et al. 2008

#### CZECH REPUBLIC, LOWER DEVONIAN



#### FRENCH-BELGIAN MUD MOUNDS, FRASNIAN



Iron-bacteria

(Siderocapsa-like, Sphaerotilus-Leptothrix-like in the internal sediments (stromatactis) of Receptaculites Rochefontaine quarry, Franchimont, Philippeville Massif

#### BALEAS GRIOTTES, SPAIN, CARBONIFEROUS



<u>Centripetal growth</u> of microstromatolitic columns (blisters and small tubes hematitized)







with endostromatolites ... and microstromatolites

#### SIERRA DEL CUERA, SPAIN, CARBONIFEROUS - PLATEAU DI ASIAGO, ITALY, JURASSIC

Infillings of original fossil cavities (here, Foraminifer)

Also Bryozoan, Gastropods, Ostracods, Tentaculids, Protoglobogerinids...



#### SIERRA DEL CUERA, SPAIN, CARBONIFEROUS

Infillings of bioperforations







## ANTI-ATLAS, MOROCCO LOWER-UPPER DEVONIAN



#### ANTI-ATLAS, MOROCCO LOWER-UPPER DEVONIAN

# Filamentous iron bacteria

Sheath

Iron encrustation (25-50% Fe)



Diameters: 1.5-4 µm [SEM]



1amet, Préat 200

#### BAJOCIAN GSSP, SAINTE-HONORINE-DES-PERTES, FRANCE

Inside a Fe-oncoid... (nucleus is a small ammonite)



# SEM observations (x1,000 ... x35,000...)

- Simple and regular filaments
- Simple filaments with regular constrictions
- Dichotomous filaments with constrictions
- Concentrations of regular sphaerules

## Diameters $\leq 2 \mu m$

with submicronic hematite Fe<sub>2</sub>0<sub>3</sub> or goethite Fe0.0H in the sheath

These morphologies are suggestive of iron bacteria

# Irregular filamentous forms (10'µm), sometimes forming a network and associated with spores



# These morphs suggest the presence of FUNGI IMPERFECTI



# Hematite/Goethite are not dispersed at random but follows regular patterns

![](_page_41_Picture_2.jpeg)

I. Today they are associated with Fe and/or Mn deposits.  $O_2$  and pH values determine the iron solubility in aqueous solutions.

2. The <u>neutrophile</u> iron bacteria are associated with the oxic/anoxic interface -Sphaerotilus, Leptothrix, Gallionella ...

3. Iron biomineralization is linked to the production of EPS - exopolymeric substances = sheaths or capsules rich in polysaccharides forming the main part of the bacterial mats. The Fe<sup>3+</sup> is <u>passively</u> precipitated in the EPS of the Recent bacterial films

![](_page_43_Figure_0.jpeg)

![](_page_44_Figure_0.jpeg)

... if iron bacteria are present, iron hydroxides are linked to oxic/anoxic gradients in poorly oxygenated waters and independent of light

![](_page_45_Figure_0.jpeg)

# Microenvironmental significance and No paleogeographical significance

# Stable isotope geochemistry of iron

- **griottes** Devonian S-France, Viséan N-Spain
- ° 'red marbles' Devonian (Frasnian), Belgium
- **Ammonitico Rosso** Jurassic, N-Italy, S-Spain, Sicily
  - ° 'red marble' Devonian, Czech Republic
  - red condensed series Devonian, Morocco
  - red lenses in slope Carboniferous, N-Spain
  - ° 'Oolite Ferrugineuse de Bayeux' mid-Jurassic Normandy

![](_page_46_Picture_8.jpeg)

Echinocardium cordatum Montacuta ferruginosa BIOLOGY

 $\bigcirc$ 

<sup>56</sup>Fe 91.76% <sup>54</sup>Fe 5.84 % <sup>57</sup>Fe 2.12 % <sup>58</sup>Fe 0.28 %

## What about Fe-isotopes?

4 outcrops (60 km)

Fe-hardgrounds Filamentous Fe-microbes Fe-microstromatolites Fe-biofilms

![](_page_47_Figure_4.jpeg)

![](_page_48_Figure_0.jpeg)

Mass-dependent isotope variations of Fe in the rock record span a range of 4 ‰ in 56Fe/54Fe ratio.

#### <sup>56</sup>Fe 91.76% <sup>54</sup>Fe 5.84 % <sup>57</sup>Fe 2.12 % <sup>58</sup>Fe 0.28

# What about Fe-isotopes?

 $\delta^{56}$ Fe = (<sup>56</sup>Fe/<sup>54</sup>Fe<sub>sample</sub>/<sup>56</sup>Fe/<sup>54</sup>Fe<sub>IRMM014</sub> - I) \* 10<sup>3</sup>

![](_page_49_Figure_3.jpeg)

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#### **EXPERIMENTS** IN BIOLOGICAL SYSTEMS

<sup>56</sup>Fe/<sup>54</sup>Fe ratios of the (solid) ferrihydrite precipitate **with an 1.5‰ increase** relative to the aqueous Fe(II) source

#### AND

Anbar EPSL 2004

the net Fe isotope fractionation is remarkably similar (1.3% more **negative**) to that obtained for dissimilatory Fe(III) reducing bacteria

SO, WHAT IN NATURAL SYSTEMS (GEOLOGY AND BIOLOGY)?

![](_page_50_Figure_0.jpeg)

# Sample preparation

A small electric drill with <u>a titanium</u> drill head was used to extract powder from individual grey and red layers (25 mg for each sample)

![](_page_51_Picture_2.jpeg)

# Analytical Procedure

- Bulk sample dissolution
- Leaching (HCI 3M for 3h at 50°C) in order to remove the carbonate phases without dissolving the other phases (silicates, oxides)

![](_page_52_Picture_3.jpeg)

![](_page_52_Picture_4.jpeg)

Separation of Fe by one-step ion-exchange chromatography

Needs iron separation and purification...

#### Iron Isotopic Analyses by Nu Plasma MC-ICP-MS

![](_page_53_Picture_1.jpeg)

Electrostatic Analyzer

Magnet

Collector block (12 Faraday cups)

Dry plasma: Aridus desolvating sample introduction system Minimalizing interferences ArOH: mass 57, ArO: mass 56, ArN: mass 54

> ... an analysis takes about 9 minutes, and consumes about 1ml of sample or 400 ng of Fe

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Cetac Aridus

# Analytical conditions

•  $\delta^{56}$ Fe = ( ${}^{56}$ Fe/ ${}^{54}$ Fe<sub>sample</sub>/ ${}^{56}$ Fe/ ${}^{54}$ Fe<sub>IRMM014</sub> -I) × 10<sup>3</sup>

 Simultaneous external normalization (Cu-doping method in dynamic mode) and standard-sample bracketing with the IRMM014 reference material;

- Cr correction on mass 54;
- Every sample in duplicate;
- Long-term accuracy and reproducibility of 0.15  $\pm$  0.06 ‰ for  $\delta^{56}$ Fe
- I sigma, n = 21) for a basalt relative to IRMM014.

Per sample: 10 days (dissolution, separation) + 2 days (ICP-MS)/3 samples Optimal timing: 15 days -1 month for 2 or 3 samples

#### **GEOCHEMICAL STUDY**

Voltascura + Forte di Campo Luserna RAI, 25 samples

CaCO<sub>3</sub>, Mg, Fe, Mn, Sr (whole rock and selective)
δ<sup>18</sup>O-δ<sup>13</sup>C (selective microdrillings)

╋

17 samples for iron isotopes (non selective and selective)

![](_page_55_Picture_5.jpeg)

Red Pink Grey

╋

#### cathodoluminescence

# The RED colour is NOT related to the iron content but to its MINERALOGICAL PHASE

![](_page_56_Figure_1.jpeg)

## $\delta^{13}$ C vs $\delta^{18}$ O RAV

![](_page_57_Figure_1.jpeg)

# FIRST CONCLUSIONS

♦ The analyzed carbonates are pure: 88-98% [CaCO<sub>3</sub>]

On relation between Mn and Fe contents ===> early diagenetic mobility of Mn confirmed by cathodoluminescence analysis (not shown here) and SEM on Mn-oncoids (Sicily) (*id.*)

◊ Very low Sr contents (50-100 ppm) ===> no aragonitic precursor?

 $\Diamond$  No meteoric influence

![](_page_59_Figure_0.jpeg)

3M HCI/3h/50°C leach + total digestion residue											
	δ56Fe ‰	∆ LEACH- ‰	RES	δ56Fe Δ ‰	LEACH-RES	∆ R-G ‰					
#100/101R-LEACH #100/101R-RES	-1,11 -1,19	0,08	#100/101G-LEAC #100/101G-RES	H -0,21 -0,09	-0,12	-0,90 <b>-1,10</b>					
#102R-LEACH #102R-RES	-0,38 <mark>-0,68</mark>	0,30	#102G-LEACH <b>#102G-RES</b>	-0,25 -0,11	-0,14	-0,13 <b>-0,57</b>					
#160R-LEACH #160R-RES	-0,02 -0,34	0,32	#160G-LEACH #160G-RES	-0,39 -0,27	-0,12	0,37 - <b>0,07</b>					
Average ∆ LEACH-RES stdev		0,23 0,13			-0,13 0,01						

+ 0.1M HCL/18h/25°C and 3M HCL/1h/50°C and 3M HCL/2h/25°C

#### VITAL EFFECT? or (BIO)FRACTIONATION FROM A MORE NEGATIVE SOURCE?

![](_page_61_Figure_1.jpeg)

#### TRY TO SOLVE FROM THE RECENT?

![](_page_62_Picture_1.jpeg)

#### Going back again to biology...

![](_page_62_Picture_3.jpeg)

LIVING MODELS

![](_page_62_Picture_5.jpeg)

IRON GRANULES IN THE BACTERIAL SHEATH (Montacuta ferruginosa)

![](_page_63_Picture_1.jpeg)

# Transmission electron microscopy

Gillan & De Ridder 2001. Chemical Geology 177:371-379

![](_page_63_Picture_4.jpeg)

Nanoparticules

![](_page_64_Picture_0.jpeg)

OUTER LAYER WITH FERRIC IRON (AMORPHOUS WITH Fe, P, Ca and TRACES of Si, Mg) 0.05-1 µm (COLLOIDAL RANGE) Infrared absortion spectra = amorphous iron oxyhydroxide gel with phoshate sorbed on the its surface rather than a pure ferric phosphate

INSOLUBLE FERRIC PHOSPHATE (E. cordatum)

![](_page_65_Figure_0.jpeg)

Table 2. Iron isotopic composition of iron-encrusted biofilms of *M. ferruginosa* (samples M1–M5) and iron-encrusted nodules and intestinal wall of *E. cordatum* (samples E1, E2, E3). First column gives the iron content (%) of the samples. Sample description in text. See Préat et al. (2008a, b) for comparison with iron isotopic composition of Rosso Ammonitico Veronese samples.

	% Fe	δ <sup>56</sup> Fe (‰)		δ <sup>57</sup> Fe (‰)		
		Avg	SD	Avg	SD	n
Bivalve M1	1.61	-0.84	0.04	-1.25	0.07	2
Bivalve M2	0.80	-0.74	0.01	-1.08	0.00	2
Bivalve M3	0.78	-1.66	0.04	-2.44	0.03	3
Bivalve M4	1.72	-1.14	0.04	-1.71	0.04	3
Bivalve M5	0.75	-1.15	0.01	-1.64	0.00	2
E1 nodule	2.73	-1.06	0.01	-1.54	0.01	2
E2 nodule	2.72	-0.76	0.06	-1.08	0.09	2
E3 wall	10.10	-1.78	0.02	-2.64	0.01	2

![](_page_66_Figure_0.jpeg)

![](_page_67_Figure_0.jpeg)

# CONCLUSIONS

- I. The red color is related to the submicronic hematite/goethite dispersed in the matrix. The hematite is a result of the activity of iron bacteria and Fungi that precipitated Fe-Mn hydroxydes at dysoxic sediment-water interfaces
- 2. The iron contents are comparable in red, pink and grey facies
- 3. The iron bacteria have passively 'fractionated' the iron isotopes at an infra-millimetric scale
- Comparison of the Fe isotopic compositions of the 'biominerals' in the Recent organisms and the red-grey Jurassic facies suggest an isotopic biofractionation of at least +0.7 ‰, and around 1‰ (or more in the Ammonitico Rosso)
- 5. The Recent sea-urchin and the bivalve thrive in similar microenvironmental conditions as the microorganisms of the Jurassic condensed red facies.

# A biosignature is probably present....

# THANK YOU

#### Future Research

# Biosignature ?

![](_page_69_Picture_3.jpeg)

Fe<sup>2+</sup>aq